# Three new iridolactone derivatives from the whole plant of Brillantaisia owariensis P. Beauv 

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#### Abstract

Owariensisone B-D, three new lactone iridoids, identified as lactone of $\alpha$-(3-acetoxymethyl-2-hydromethylcyclopent-3-en-yl) ethanoic acid (1), lactone of $\alpha$-(4- $\beta$-acetoxy-3- $\beta$ -acetoxymethyl-2-hydromethyl-3-hydroxycyclopentyl) ethanoic acid (2) and lactone of $\alpha-(4-\beta$ -acetoxy-3- $\alpha$-acetoxymethyl-2-hydromethyl-3-hydroxycyclopentyl) ethanoic acid (3) were isolated from the $n$-hexane extract of the whole plant of Brillantaisia owariensis. Their structures were established by interpretation of their spectral data, mainly ESI-HRMS(TOF), 1D-NMR ( ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$ ) and 2D-NMR ( ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY, HSQC, HMBC and ROESY) and by comparison with the literature data.


Keywords: Brillantaisia owariensis; Acanthaceae; Lactone iridoids; Owariensisone.

## Introduction

The genus Brillantaisia belongs to the family Acanthaceae and is found growing throughout tropical Africa and Madagascar and occupies a prominent position in traditional medicine (Ngbolua et al., 2013). Eleven species are known in Cameroon, and among these, Brillantaisia owariensis P. Beauv (Synonym Brillantaisia patula T. Anderson), a large erect shrub with a purple-blue flowers (Heine, 1963). It is found growing in Nigeria, Togo, West Cameroon and across Uganda and Angola. The leaves are used for the treatment of anaemia (Ngbolua et al., 2013), rheumatism, menstrual pain, stomach ache and for their antiplasmodial and analgesic potentials (Asai et al., 2012; Makambila-Koubemba et al., 2011; Mbatchi et al., 2006). Previous study indicated that the alcoholic extract has antibacterial and antioxidant activities (Aluko et al., 2014; Faparusi et al., 2012,) but the phytochemical investigations were limited. Our previous contribution reported the isolation of one new lactone iridoid in the methanol extract with flavonoid and glycosides (Foning Tebou et al., 2016). In an extension of our studies, the $n$-hexane soluble part of the methanol extract of the whole plant of $B$. owariensis was examined in greater detail and three novel lactone iridoids (1-3) were isolated in minor concentration.

## 2. Results and discussion

Purification of the $n$-hexane soluble fraction of the crude MeOH extract afforded three new compounds, owariensisone $\mathrm{B}, \mathrm{C}$ and D (1-3) (Fig. 1).

Compound 1 was obtained as yellowish gum. Its molecular formula was determined as $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{5}$ on the basis of its ESI-HRMS(TOF) spectrum, exhibiting a pseudo-molecular ion
peak at $m / z 249.0734[\mathrm{M}+\mathrm{Na}]^{+}$(calcd. for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{5} \mathrm{Na}$. 249.0739) with five degrees of unsaturation. Its ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum (Table 1) shows one olefinic proton at $\delta_{\mathrm{H}} 5.83(\mathrm{H}-7$, brqt, $J=1.6 \mathrm{~Hz})$, two pairs of gem-coupled methylene protons at $\delta_{\mathrm{H}} 2.60$ and $2.65(\mathrm{H}-6 \alpha$ and $\mathrm{H}-$ $6 \beta$ ), and at $\delta_{\mathrm{H}} 2.80$ and $2.83(\mathrm{H}-4 \alpha$ and $\mathrm{H}-4 \beta$, each d, $J=14.6 \mathrm{~Hz}$ ), two pairs of oxymethylene protons at $\delta_{\mathrm{H}} 4.59$ and $4.72(\mathrm{H}-10 \alpha$ and $\mathrm{H}-10 \beta$, each d, $J=13.2 \mathrm{~Hz})$ and $\delta_{\mathrm{H}} 4.26$ and $4.58(\mathrm{H}-$ $1 \alpha$ and $\mathrm{H}-1 \beta$ ), and one methine proton at $\delta_{\mathrm{H}} 2.99(\mathrm{H}-9, \mathrm{dd}, J=6.5,4.9 \mathrm{~Hz})$ attributed to the boschnialactone type skeleton (Sakan et al., 1967; Sisido et al., 1968; Callant et al., 1983; Tanaka et al., 1993; Hilgraf et al., 2012). The signal observed at $\delta_{\mathrm{H}} 2.09$ (3H, H-2', s) indicated the presence of an acetyl group in this compound. This was supported by the ${ }^{13} \mathrm{C}$ NMR spectrum exhibiting eleven carbon signals including two ester carbonyls at $\delta_{\mathrm{C}} 174.4$ (C3 ) and 172.4 (C-1'), an acetoxymethyl carbon at $\delta_{\mathrm{C}} 20.7$ (C-2'), two ethylenic carbons at $\delta_{\mathrm{C}}$ 136.7 (C-8) and 131.1 (C-7), four methylene carbons at $\delta_{\mathrm{C}} 68.4$ (C-1), 62.5 (C-10), 48.9 (C-6) and 44.4 (C-4), one methine carbon at $\delta_{\mathrm{C}} 55.4$ (C-9) and one quaternary carbon at $\delta_{\mathrm{C}} 78.9$ (C5) bearing an hydroxy group. These NMR data are very closed to owariensisone except for a supplementary acetyl group in C-10 position (Foning Tebou et al., 2016). The ${ }^{1} J_{\mathrm{C}-\mathrm{H}}$ correlation in the HSQC spectrum allowed us to attribute to each carbon the corresponding proton. Thus, the two methylene protons at $\delta_{\mathrm{H}} 2.60$ and 2.65 (H-6) were linked to carbon C-6 and those at $\delta_{\mathrm{H}} 2.80$ and 2.83 (H-4) were correlated to carbone C-4, and the two pairs oxymethylene protons group at $\delta_{\mathrm{H}} 4.26$ and $4.58(\mathrm{H}-1)$ and 4.59 and $4.72(\mathrm{H}-10)$ were linked to carbon C-1 and C-10, respectively. In the COSY spectrum, protons at $\delta_{\mathrm{H}} 2.60$ and $2.65(\mathrm{H}-$ 6) were correlated with proton at $\delta_{\mathrm{H}} 5.83$ (H-7). Mains correlations were also observed between the protons at $\delta_{\mathrm{H}} 4.26$ and $4.58(\mathrm{H}-1)$ and proton at $\delta_{\mathrm{H}} 2.99(\mathrm{H}-9)$.

The two gem-methylene systems $\mathrm{H}-1$ and $\mathrm{H}-4$ were allocated to the $\delta$-lactone group by the HMBC cross peaks with the carbonyl carbon at $\delta_{\mathrm{C}} 174.4$ thus confirming its $\mathrm{C}-3$ position. From the HMBC spectrum, correlations observed between the methylene protons $\mathrm{H}-10$ and carbons C-7, C-8 and C-9 suggested that this oxymethylene was located at C-8. The carbonyl of the methyl ester at 172.4 (C-1') was assigned to $\mathrm{C}-10$ by long-range correlation between protons at $\delta_{\mathrm{H}} 2.09\left(\mathrm{H}_{3}-2^{\prime}\right)$ and carbon at $\delta_{\mathrm{C}} 62.5(\mathrm{C}-10)$.

The $\beta$-orientation of hydrogen at C-9 and hydroxyl group at C-5 was supported by the biosynthetic pathway to iridomyrmercin, isoiridomyrmercin and owariensisone, respectively (Lunn, 1961; Foning Tebou et al., 2016), as in usual iridoids. This is confirmed by the coupling constants of H-9 from 6.5 and 4.9 Hz with both protons $\mathrm{H}-1$, characteristic of pseudo-axial-axial and pseudo-axial-equatorial system as in owariensisone. On the basis of afore mentioned information, the structure of compound 1 was elucidated as $10-O$-acetyl
owariensisone or lactone of $\alpha$-(3-acetoxymethyl-2-hydromethylcyclopent-3-en-yl) ethanoic acid named owariensisone B (Fig. 1).

Compound $\mathbf{2}$ was obtained as yellowish gum. Its molecular formula was determined as $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{8}$ on the basis of its pseudo-molecular ion peak at $\mathrm{m} / \mathrm{z} 325.0905[\mathrm{M}+\mathrm{Na}]^{+}$(calcd. for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{8} \mathrm{Na} 325.0899$ ) in the ESI-HRMS(TOF) spectrum, with five degrees of unsaturation. The ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum (Table 1) shows signals for two acetyl groups [ $\delta_{\mathrm{H}} 2.09$ (3H, H-2', s), 2.07 (3H, H-2'’, s); $\delta_{\mathrm{C}} 172.5$ (C-1'), 171.8 (C-1'’), 20.7 (C-2'), 21.0 (C-2’’)] and signals of iridoid lactone: two pairs of methylene protons at $\delta_{\mathrm{H}} 2.00$ and 2.43 (H-6 $\alpha, \mathrm{dd}, J$ $=14.9,1.8 \mathrm{~Hz}$ and $\mathrm{H}-6 \beta$, dd, $J=14.9,5.0 \mathrm{~Hz})$ and $\delta_{\mathrm{H}} 2.73$ and $2.85(\mathrm{H}-4 \alpha, \mathrm{dd}, J=14.8,0.8$ Hz and $\mathrm{H}-4 \beta, \mathrm{~d}, J=14.8 \mathrm{~Hz}$ ), two pairs of oxymethylene protons at $\delta_{\mathrm{H}} 4.49$ and $4.53(\mathrm{H}-1 \alpha$, dd, $J=11.9,7.7 \mathrm{~Hz}$ and $\mathrm{H}-1 \beta$, dd, $J=11.9,10.0 \mathrm{~Hz})$ and $\delta_{\mathrm{H}} 4.12$ and $4.38(\mathrm{H}-10 \alpha, \mathrm{~d}, J=11.6$ Hz and $\mathrm{H}-10 \beta, \mathrm{~d}, J=11.6 \mathrm{~Hz}$ ), one methine proton at $\delta_{\mathrm{H}} 2.57(\mathrm{H}-9, \mathrm{dd}, J=10.0,7.7 \mathrm{~Hz})$ and one oxymethine proton at $\delta_{\mathrm{H}} 5.07(\mathrm{H}-7, \mathrm{dd}, J=5.0,1.8 \mathrm{~Hz})$. This was supported by the ${ }^{13} \mathrm{C}-$ NMR spectrum exhibiting signals of an ester carbonyl at $\delta_{\mathrm{C}} 174.6$ (C-3), four methylene carbons at $\delta_{\mathrm{C}} 66.6$ (C-1), 67.4 (C-10), 45.3 (C-6) and 44.9 (C-4), two methine carbons at $\delta_{\mathrm{C}}$ 80.9 (C-7) and 51.9 (C-9) and two quaternary carbons at $\delta_{\mathrm{C}} 82.1$ (C-8) and 80.1 (C-5) bearing each an hydroxy group (Table 1). Their attributions were assigned by analysis of COSY and ${ }^{1} J_{\mathrm{C}-\mathrm{H}} \mathrm{HSQC}$ spectra. In the COSY spectrum, proton at $\delta_{\mathrm{H}} 5.07$ (H-7) was correlated with protons at $\delta_{\mathrm{H}} 2.00$ and 2.43 (H-6). Another correlation was also observed between the protons at $\delta_{\mathrm{H}} 4.49$ and $4.53(\mathrm{H}-1)$ and proton at $\delta_{\mathrm{H}} 2.57(\mathrm{H}-9)$. As compared to compound 1, the ethylenic group in $\Delta_{7,8}$ was replaced by two oxycarbones at C-7 and C-8 as in patriscabrol (Kouna et al., 1994).

Analysis of HMBC spectrum showed correlations between the protons at $\delta_{\mathrm{H}} 4.12$ and 4.38 ( $\mathrm{H}-$ $10)$ and carbons at $\delta_{\mathrm{C}} 82.1$ (C-8), 80.9 (C-7) and $67.4(\mathrm{C}-9)$ suggesting that this oxymethylene was located at $\mathrm{C}-8$ as in $\mathbf{1}$. The carbonyl of the methyl ester at $172.5\left(\mathrm{C}-1^{\prime}\right)$ was assigned to C 10 by long-range correlation with protons $\mathrm{H}-10$, while the one at 171.8 ( $\mathrm{C}-1^{\prime \prime}$ ) was linked to $\mathrm{C}-7$ by the long-range correlation with proton at $\delta_{\mathrm{H}} 5.07(\mathrm{H}-7)$. The position of the lactone at $\delta_{\mathrm{C}} 174.6$ was determined as in $\mathbf{1}$. The $\beta$-axial orientation of $\mathrm{H}-9$ was deduced from the coupling constant of 10.0 and 7.7 Hz with protons $\mathrm{H}-1 \quad \alpha$-axial and $\mathrm{H}-1 \quad \beta$-equatorial, respectively. The $\alpha$-equatorial orientation of H-7 was deduced from the coupling constant of 5.0 and 1.8 Hz with $\mathrm{H}-6 \alpha$-axial and H-6 $\beta$-equatorial, respectively as observed for similar protons in jioglutoside (Morota et al., 1989) and patriscabrol (Kouna et al., 1994). Thus, the hydroxyl at C - 7 is $\beta$-oriented. As observed in jioglutolide the $\delta$-lactone ring has a boat conformation (1B4) (Morota et al., 1989), as observed by the ROESY correlations between

H-1 and H-4 $\alpha$-axial. Correlations observed in the ROESY spectrum between $\mathrm{H}-9$ and $\mathrm{H}-10 \beta$, indicated a $\beta$-axial orientation of $\mathrm{H}-10$ and an $\alpha$-axial orientation of hydroxyl at $\mathrm{C}-8$ position (Fig. 2). On the basis of afore mentioned information, the structure of compound 2 was elucidated as lactone of $\alpha$-(4- $\beta$-acetoxy-3- $\beta$-acetoxymethyl-2-hydromethyl-3hydroxycyclopentyl) ethanoic acid named owariensisone $\mathbf{C}$ (Fig.1).

Compound 3 was obtained as yellowish gum. It has the same molecular formula $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{8}$ as compound 2, and differs from the latter only by the orientation of C-8 oxymethylene. Its ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum (Table 1) was so closed to that of $\mathbf{2}$ and its ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum is almost superimposable on that of $\mathbf{2}$, except for the signal of oxymethylene $\mathrm{C}-10$ $\delta_{\mathrm{C}} 65.0(\Delta-2.4 \mathrm{ppm})$ and methine C-9 $\delta_{\mathrm{C}} 54.0(\Delta+2.1 \mathrm{ppm})$. From the analysis of HMBC spectrum, the oxymethylene, the carbonyl of the methyl ester and the position of the lactone have been solved as in 2. The absence of correlation between $\mathrm{H}-10$ and $\mathrm{H}-9$ in the ROESY spectrum indicated the $\alpha$-axial orientation of C -10 (Fig.2). On the basis of afore mentioned information, the structure of compound 3 was elucidated as lactone of $\alpha-(4-\beta$-acetoxy- $3-\alpha-$ acetoxymethyl-2-hydromethyl-3-hydroxycyclopentyl) ethanoic acid named owariensisone D (Fig.1).

## 3. Experimental

### 3.1. General

IR spectra were recorded with a Shimadzu FT-IR-8400S (Shimadzu, France) spectrophotometer. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$-NMR spectra were recorded on a Bruker Avance III 600 spectrometer equipped with a cryoprobe ( ${ }^{1} \mathrm{H}$ at 600 MHz and ${ }^{13} \mathrm{C}$ at 150 MHz ). 2D NMR experiments were recorded by means of standard Bruker microprograms (Xwin-NMR version 2.1 software TopSpin 3.2). Chemical shifts ( $\delta$ ) are reported in parts per million ( ppm ) using the residual solvent signals as secondary reference relatively to TMS $(\delta=0)$, while the coupling constants ( $J$ values) are given in Hertz (Hz). ESI-MS(TOF) and ESI-HRMS(TOF) spectra were recorded using a Micromass Q-TOF micro instrument (Manchester, UK) equipped with an electrospray source. The samples were introduced by direct infusion in a solution of MeOH at a rate of $5 \mu \mathrm{~L} \mathrm{~min}^{-1}$. The optical rotations were measured on a Bellingham \& Stanley ADP 220 polarimeter (Bellingham + Stanley Ltd., United Kingdom). Column chromatography was run on Merck silica gel 60 (70-230 mesh) and gel permeation on Sephadex LH-20 while TLC was carried out on silica gel $\mathrm{GF}_{254}$ pre-coated plates with detection accomplished by spraying with $50 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ followed by heating at $100^{\circ} \mathrm{C}$, or by visual inspection under UV lamp at 254 and 365 nm .

### 3.1. Plant material

The whole plant of B. owariensis was collected at Tsinkop, Menoua Division, West Region of Cameroon, in October 2013. Authentication was done by Mr Victor Nana, a botanist of the Cameroon National Herbarium, Yaoundé, where the voucher specimen ( $\mathrm{N}^{0} 34376 / \mathrm{HNC}$ ) has been deposited.

### 3.2. Extraction and isolation

The air-dried and plant material ( 4 Kg ) was powdered and extracted at room temperature with methanol ( 3 x 15L, 72 h ). The solvent was evaporated under reduced pressure, leaving an extract ( 110 g ). Part of this extract ( 100 g ) was extracted with $n$-hexane yielding 30 g fraction after evaporation to dryness. Part of $n$-hexane-soluble fraction $(\mathbf{2 5} \mathrm{g})$ was subjected to silica gel ( $0,200-0,500 \mathrm{~mm}$ ) column chromatography ( $40 \times 800 \mathrm{~mm}$ ) using Hex-EtOAc (90:10 $\rightarrow 0: 100$ ) gradient graduated elution. Many fractions of 250 mL each were collected and combined on the basis of their TLC profiles to give 3 fractions: A, B and C. Fraction B ( 5.1 g ) was purified on silica gel $(0,063-0,200 \mathrm{~mm})$ column chromatography $(30 \times 600 \mathrm{~mm})$ using an isochratic eluent system Hex-EtOAc (50:50). Fractions of 10 mL were collected and combined on the basis of TLC profiles to give 3 sub-fractions $B_{1}(1.5 \mathrm{~g}), \mathrm{B}_{2}(102 \mathrm{mg})$ and $\mathrm{B}_{3}$ $(1 \mathrm{~g})$. Sub-fraction $\mathrm{B}_{2}(102 \mathrm{mg})$ was subjected to silica gel $(0,063-0,200 \mathrm{~mm})$ column chromatography ( $10 \times 200 \mathrm{~mm}$ ) using Hex-EtOAc ( $60: 40$ ) as eluent to give compound 1 (8 mg ) and a mixture of $\mathbf{2}$ and $\mathbf{3}(50 \mathrm{mg})$. This mixture was combined to sub-fractions $B_{1}$ and $B_{3}$ and purified by chromatography on silica gel ( $0,063-0,200 \mathrm{~mm}$ ) column chromatography ( 20 $\times 400 \mathrm{~mm}$ ) using the same system solvent to yield compounds $\mathbf{3}(12 \mathrm{mg})$ and $\mathbf{2}(9 \mathrm{mg})$.

## New compounds

Owariensisone B: Yellowish gum; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data, see Table $1 ;[\alpha]_{\mathrm{D}}{ }^{21}-5.2(\mathrm{MeOH} c$ $0.06)$; IR ( NaCl ) $v_{\max }\left(\mathrm{cm}^{-1}\right) 3350-3300,1070,1040(\mathrm{OH}), 1643(\mathrm{C}=\mathrm{O}), 1605,1580,1520$ (C=C aromatic), 1650 (C-O); ESI-HRMS(TOF) (positive ion mode) $m / z: 249.0734[\mathrm{M}+\mathrm{Na}]^{+}$ (calcd. for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{5} \mathrm{Na} 249.0739$ ).
Owariensisone C: Yellowish gum; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data, see Table $1 ;[\alpha]_{\mathrm{D}}{ }^{21}-20(\mathrm{MeOH} c$ $0.05)$; IR $(\mathrm{NaCl}) v_{\max }\left(\mathrm{cm}^{-1}\right) 3350-3300,1070,1040(\mathrm{OH}), 1643(\mathrm{C}=\mathrm{O}), 1605,1580,1520$ (C=C aromatic), 1650 (C-O); ESI-HRMS(TOF) (positive ion mode) m/z: $325.0905[\mathrm{M}+\mathrm{Na}]^{+}$ (calcd. for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{8} \mathrm{Na} 325.0899$ ).
Owariensisone D: Yellowish gum; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{CNMR}$ data, see Table 1; $[\alpha]_{\mathrm{D}}{ }^{21}-87.5(\mathrm{MeOH} c$ $0.02)$; IR $(\mathrm{NaCl}) v_{\max }\left(\mathrm{cm}^{-1}\right) 3350-3300,1070,1040(\mathrm{OH}), 1643(\mathrm{C}=\mathrm{O}), 1605,1580,1520$
(C=C aromatic), 1650 (C-O); ESI-HRMS(TOF) (positive ion mode) m/z: $325.0892[\mathrm{M}+\mathrm{Na}]^{+}$ (calcd. for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{8} \mathrm{Na} 325.0899$ ).

## Acknowledgments

The authors are grateful to the University of Dschang for financing some consumables used in this work, to the "Service Commun d'Analyses" and "Groupe Isolement et Structure", to the "Institut de Chimie Moléculaire de Reims" for the spectroscopic and spectrometric analysis on the ESIMS et NMR equipement of the PlAnet Platform. The EU-programme FEDER to the PlAneT CPER project is gratefully acknowledged.

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1


2


3

Fig.1. Structures of compounds 1-3 isolated from the $n$-hexane soluble extract of Brillantaisia owariensis.


2


3

Fig.2. ROESY effects in compounds 2-3

Table 1: ${ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz})$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz})$ data of compounds $\mathbf{1 - 3}$ in $\mathrm{CD}_{3} \mathrm{OD}$.

| $\mathrm{N}^{\circ}$ | $\mathbf{1}$ |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
|  | ${ }^{13} \mathrm{C}$ | ${ }^{1} \mathrm{H}(\mathrm{J}$ in Hz$)$ | ${ }^{13} \mathrm{C}$ | ${ }^{1} \mathrm{H}(J$ in Hz $)$ | $\mathbf{3}$ |  |
| $\mathbf{1}$ | 68.4 | $4.26(1 \mathrm{H}, \mathrm{dd}, 11.9,6.5)$ | 66.6 | $4.49(1 \mathrm{H}, \mathrm{dd}, 11.9,7.7)$ | 66.9 | $4.21(1 \mathrm{H}, \mathrm{t}, 11.2)$ |
| $\mathbf{3}$ | 174.4 | $4.58(1 \mathrm{H}, \mathrm{dd}, 11.9,4.9)$ |  | $4.53(1 \mathrm{H}, \mathrm{dd}, 11.9,10.0)$ |  | $4.57(1 \mathrm{H}, \mathrm{dd}, 11.6,6.2)$ |
| $\mathbf{4}$ | 44.4 | $2.80(1 \mathrm{H}, \mathrm{d}, 14.6)$ | 174.6 |  | 174.5 |  |
|  |  | $2.83(1 \mathrm{H}, \mathrm{d}, 14.6)$ |  | $2.73(1 \mathrm{H}, \mathrm{dd}, 14.8,0.8)$ | 44.7 | $2.57(1 \mathrm{H}, \mathrm{d}, 15.1)$ |
| $\mathbf{5}$ | 78.9 |  | 80.1 |  |  | $2.78(1 \mathrm{H}, \mathrm{d}, 15.1)$ |
| $\mathbf{6}$ | 48.9 | $2.60(1 \mathrm{H}, \mathrm{dq}, 18.1,1.0)$ | 45.3 | $2.00(1 \mathrm{H}, \mathrm{dd}, 14.9,1.8)$ | 46.1 | $1.91(1 \mathrm{H}, \mathrm{dd}, 14.5,1.6)$ |
|  |  | $2.65(1 \mathrm{H}, \mathrm{dm}, 18.1)$ |  | $2.43(1 \mathrm{H}, \mathrm{dd}, 14.9,5.0)$ |  | $2.46(1 \mathrm{H}, \mathrm{dd}, 14.5,5.0)$ |
| $\mathbf{7}$ | 131.1 | $5.83(1 \mathrm{H}, \mathrm{brqt}, 1.6)$ | 80.9 | $5.07(1 \mathrm{H}, \mathrm{dd}, 5.0,1.8)$ | 80.8 | $5.17(1 \mathrm{H}, \mathrm{dt}, 5.0,1.6)$ |
| $\mathbf{8}$ | 136.7 |  | 82.1 |  | 82.1 |  |
| $\mathbf{9}$ | 55.4 | $2.99(1 \mathrm{H}, \mathrm{dd}, 6.5,4.9)$ | 51.9 | $2.57(1 \mathrm{H}, \mathrm{dd}, 10.0,7.7)$ | 54.0 | $2.45(1 \mathrm{H}, \mathrm{dd}, 11.1,6.2)$ |
| $\mathbf{1 0}$ | 62.5 | $4.59(1 \mathrm{H}, \mathrm{d}, 13.2)$ | 67.4 | $4.12(1 \mathrm{H}, \mathrm{d}, 11.6)$ | 65.0 | $3.91(1 \mathrm{H}, \mathrm{d}, 11.6)$ |
|  |  | $4.72(1 \mathrm{H}, \mathrm{d}, 13.2)$ |  | $4.38(1 \mathrm{H}, \mathrm{d}, 11.6)$ |  | $4.18(1 \mathrm{H}, \mathrm{d}, 11.6)$ |
| $\mathbf{1 1}$ | 172.4 |  | 172.5 |  | 172.4 |  |
| $\mathbf{1 2}$ | 20.7 | $2.09(3 \mathrm{H}, \mathrm{s})$ | 20.7 | $2.09(3 \mathrm{H}, \mathrm{s})$ | 20.7 | $2.09(3 \mathrm{H}, \mathrm{s})$ |
| $\mathbf{1 3}$ |  |  | 171.8 |  | 171.2 |  |
| $\mathbf{1 4}$ |  |  | 21.0 | $2.07(3 \mathrm{H}, \mathrm{s})$ | 21.0 | $2.07(3 \mathrm{H}, \mathrm{s})$ |


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