

Complex secondary metabolites from Ludwigia leptocarpa with potent antibacterial and antioxidant activities

Florence Déclaire Mabou, Jean-De-Dieu Tamokou, David Ngnokam, Laurence Voutquenne-Nazabadioko, Jules-Roger Kuiate, Prasanta Kumar Bag

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1 Original Article

3	Complex secondary metabolites from Ludwigia leptocarpa with potent antibacterial and								
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6	Florence Déclaire Mabou ¹ , Jean-de-Dieu Tamokou ^{2*} , David Ngnokam ^{1**} , Laurence								
7	Voutquenne-Nazabadioko ³ , Jules-Roger Kuiate ² , Prasanta Kumar Bag ⁴								
8	i outqueinie Mazuelanono, vares Moger Maine , Masania Maina Zug								
9	¹ Laboratory of Environmental and Applied Chemistry, Department of Chemistry, Faculty of								
10	Science, University of Dschang, PO. Box 67 Dschang, Cameroon.								
11	³ Laboratory of Microbiology and Antimicrobial Substances, Department of Biochemistry,								
12	Faculty of Science, University of Dschang, PO. Box 67 Dschang, Cameroon.								
13	³ Groupe Isolement et Structure, Institut de Chimie Moléculaire de Reims (ICMR), CNRS								
14	UMR 7312, Bat. 18 BP.1039, 51687 Reims cedex 2, France.								
15	⁴ Department of Biochemistry, University of Calcutta, 35 Ballygunge Circular Road, Kolkata-								
16	700 019, India.								
17									
18	Abbreviated Names:								
19	Mabou FD, Tamokou J-d-D, Ngnokam D, Voutquenne-Nazabadioko L, Kuiate JR, Bag PK								
20 21	Address Correspondence to:								
22	*Dr Jean-de-Dieu Tamokou, Laboratory of Microbiology and Antimicrobial Substances,								
23	Department of Biochemistry, Faculty of Science, University of Dschang, P O. Box 67								

- Dschang, Cameroon; E-mail: jtamokou@yahoo.fr / jean.tamokou@univ-dschang.org; Tel:
 +237 677 000 897.
- **Professor David Ngnokam, Laboratory of Environmental and Applied Chemistry,
 Department of Chemistry, Faculty of Science, University of Dschang, P O. Box 67, Dschang,
 Cameroon; E-mail: dngnokam@yahoo.fr; Tel: +237 696 710 992.
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30 **Conflict of Interest:**

- 31 There is no conflict of interest to disclose.
- 32

33 Summary

34 Diarrhea continues to be one of the most common causes of morbidity and mortality among infants and children in developing countries. The present study aims to evaluate the 35 antibacterial and antioxidant activities of extracts and compounds from *Ludwigia leptocarpa*, 36 a plant traditionally used for its vermifugal, anti-dysenteric and antimicrobial properties. The 37 methanol extract was prepared by maceration from dried whole plant and successively 38 39 extracted with ethyl acetate and *n*-butanol to obtain EtOAc and *n*-BuOH extracts respectively. The column chromatography of the EtOAc and *n*-BuOH extracts followed by purification of 40 different fractions led to the isolation of ten known compounds. Structures of isolated 41 compounds were assigned on the basis of spectra analysis, and by comparison with those from 42 the literature. The antioxidant activity was evaluated by the 1,1-diphenyl-2-picrylhydrazyl 43 (DPPH) and gallic acid equivalent antioxidant capacity (GAEAC) assays. The antibacterial 44 45 activity was assessed by performing minimum inhibitory concentration (MIC), minimum bactericidal concentration (MBC) against the strains of Gram-positive bacterium, 46 47 Staphylococcus aureus (a major cause of community and hospital-associated infection), and Gram-negative multi-drug resistant bacteria, Vibrio cholerae (causative agent of cholera) and 48 Shigella flexneri (causative agent of shigellosis). All of the extracts showed different degrees 49 of antioxidant and antibacterial activities. 2β -hydroxyoleanolic acid, (2R, 3S, 2''S)-50 3",4',4",5,5",7,7"-heptahydroxy-3,8"-biflavanone and luteolin-8-C-glucoside displayed 51 the largest antibacterial and antioxidant properties which were in some cases equal or higher 52 53 than those of reference drugs. The overall results of the present study shows that L. leptocarpa has potentials as a source of natural anti-diarrhoeal and anti-free radical products, given 54 further investigations. 55

56 *Keywords*: *Ludwigia leptocarpa*, Onagraceae, triterpenoids, flavonoids, antibacterial,
57 antioxidant

59 **1. Introduction**

60 In developing countries, particularly in Africa, poor sanitation exposed the people to a wider array of microbial pathogens which increases their susceptibility to bacterial infections 61 (1). It is reported that each year, 300,0000 children die of diarrhoeal diseases. Among the 62 diarrhoeal diseases, cholera is one of the most important causal elements in terms of severity 63 of the disease and outcomes. Several epidemics of cholera have been reported from different 64 parts of Cameroon and abroad (2-5). Vibrio cholerae strains belonging to O1 and O139 65 66 serogroups are responsible to cause cholera in the form of epidemics and pandemics (6,7). In recent years, reported cholera cases increased steadily reaching more than 300' 000 cases 67 including more than 7'500 deaths during 2010 (8). As populations of poor countries continue 68 to coalesce in mega-cities with low levels of sanitation and people move rapidly around the 69 globe, new and more virulent strains of V. cholerae are expected to disseminate more rapidly 70 (9,10). This makes cholera one of the most rapidly fatal infectious illnesses known. 71

The continuous emergence of multidrug-resistant (MDR) Vibrio cholerae strains 72 drastically, reduces the efficacy of our antibiotic armory and, consequently, increases the 73 frequency of the rapeutic failure (11, 12). In many regions, affected by this pathogen, local and 74 75 indigenous plants are often the only available means of treating such infections. Among the known plant species on earth (estimated at 250,000–500,000) only a small fraction have been 76 77 investigated for the presence of antimicrobial compounds and only 1-10% of plants are used 78 by humans (13,14). Plant natural products can also have antioxidant potential. These include 79 phenolic compounds, alkaloids, terpenoids and essential oils. Plant based antioxidant compounds (15) play a defensive role by preventing the generation of free radicals and hence 80 81 are extremely beneficial to alleviate infectious diseases generating free radicals as well as 82 diseases caused by oxidative stress such as cardiovascular diseases, diabetes, inflammation, degenerative diseases, cancer, anemia, and ischemia (16). 83

Ludwigia leptocarpa (Nutt) Hara (Onagraceae or Oenotheraceae) is a herbaceous plant 84 species that is also well represented in North America and in tropical Africa (17). In 85 traditional medicine in Nigeria, an infusion of the plant is part of a mixture to treat 86 rheumatism (18). A leaf infusion has laxative, vermifugal and anti-dysenteric properties. 87 Previous works on this genus have revealed the presence of flavonoids (19,20), cerebrosides 88 and triterpenoids (20,21). It has recently been reported that alcoholic extracts of the leaves of 89 L. octovalvis, L. abyssinica and L. decurrens have potential antioxidant, antibacterial and 90 antifungal activities (22,23). To the best of our knowledge, there has been no documented 91 report on the antioxidant and antibacterial properties of *L. leptocarpa* against diarrhea strains. 92 Hence, the aim of this study was to investigate the antibacterial and antioxidant properties of 93 extracts and compounds from L. leptocarpa. 94

95

96 2. Materials and methods

97 *2.1. Experimental*

The IR spectra were recorded with a Shimadzu FT-IR-8400S (Shimadzu, France) 98 spectrophotometer. ¹H (500 MHz) and ¹³C (125 MHz) Nuclear Magnetic Resonance (NMR) 99 spectra were recorded on a BRUKER Avance DRX-500 spectrometer (Bruker, 100 Wissembourg, France) equipped with a BBFO + 5 mm sonde. ¹H (600 MHz) and ¹³C 101 (150 MHz) NMR spectra were recorded on a BRUKER Avance III-600 spectrometer 102 (Bruker, Wissembourg, France) equipped with a cryoplatform using CD₃OD with 103 tetramethylsilane (TMS) as the internal standard. Time of flight electrospray ionization 104 105 mass spectrometry (TOF-ESIMS) and High-resolution time of flight electrospray ionization mass spectrometry (HR-TOFESIMS) experiments were performed using a Micromass Q-106 TOF micro instrument (Manchester, UK) with an electrospray source. The samples were 107 introduced by direct infusion in a solution of methanol (MeOH) at a rate of 5 µL min⁻¹. 108

109 Column chromatography was run on Merck silica gel (VWR, France) 60 (70-230 110 mesh) and gel permeation on Sephadex LH-20 (VWR, France), while thin layer 111 chromatography (TLC) was carried out on silica gel GF254 pre-coated plates with 112 detection accomplished by spraying with 50% H₂SO₄ followed by heating at 100 °C or 113 by visualizing with a ultra-violet (UV) lamp at 254 and 365 nm.

- 114
- 115 2.2. Plant material

The whole plant of *L. leptocarpa* was collected in Foto village (Menoua Division,
Western region of Cameroon), in April 2011. Authentication was performed by Victor Nana,
a botanist of the Cameroon National Herbarium, Yaoundé, where a voucher specimen (N°
38782/HNC) has been deposited.

120

121 2.3. Extraction and isolation

The dried whole plant of *L. leptocarpa* (4 kg) was extracted with MeOH at room temperature for 3 days, and the extract was concentrated to dryness under reduced pressure to yield a dark crude extract (102 g). Part of residue obtained (97 g) was suspended in water (200 mL) and successively extracted with ethyl acetate (EtOAc) and *n*-butanol (*n*-BuOH) which were concentrated to dryness under reduced pressure to afford EtOAc (20 g) and *n*-BuOH (40 g) extracts, respectively.

According to the antimicrobial and antioxidant assays, the EtOAc and *n*-BuOH extracts were submitted to further separation and purification. Part of EtOAc extract (15 g) was purified over silica gel column eluted with hexane containing increasing EtOAc (10%, 20%, 30%, 40%, 50%, 60%, 70% and 80%) and with EtOAc containing increasing MeOH (10% and 20%). Six fractions were obtained A, B, C, D, E and F. Fraction D (1.7 g) was purified over silica gel column eluted with the mixture hexane-EtOAc (7:3) to give compounds **1** and

2 (17 mg and 22 mg, respectively). Fraction E (3.1 g) was purified over silica gel column 134 chromatography eluted with the mixture hexane-EtOAc (6:4) to give compound 3 (17 mg). 135 Part of *n*-BuOH extract (30 g) was purified over silica gel column chromatography, eluted 136 137 with EtOAc containing increasing MeOH (10%, 20%, 30%, 40% and 50%). Five fractions (G_1 - G_5) were obtained. Fraction G_1 (2.5 g) was purified over silica gel column eluted 138 with EtOAc to give the compounds 4 (19 mg) and 5 (16 mg). Fraction $G_2(3.1 \text{ g})$ was purified 139 over silica gel column chromatography eluted with the mixture EtOAc-MeOH (8.5:1.5) to 140 give compounds 5 (25 mg) and 6 (13 mg). Fractions G_3 and G_4 (5.4 g) were combined and 141 purified by silica gel column chromatography eluting with the mixture of EtOAc-MeOH-H₂O 142 (8:1:1) to give the compounds 7 (38 mg) and 8 (24 mg). Fraction G₅ (2.5 g) was purified by 143 silica gel column chromatography eluting with the mixture of EtOAc-MeOH-H₂O (7:2:1) to 144 give the compounds 9 (66 mg) and 10 (40 mg). 145

- 146 Oleanolic acid (1): white amorphous powder from hexane-EtOAc; $C_{30}H_{48}O_3$.
- 147 2β -hydroxyoleanolic acid (2): white amorphous powder from hexane-EtOAc; C₃₀H₄₈O₄.
- 148 (2R,3S,2"S)-3",4',4",5,5",7,7"-heptahydroxy-3,8"-biflavanone (**3**): white amorphous 149 powder from hexane-EtOAc; C₃₀H₂₂O₁₁; high resolution electron impact mass spectrometry 150 (HRESIMS, positive-ion mode) m/z: 581.1057 [M + Na]⁺(calcd. for C₃₀H₂₂O₁₁Na 151 :581.1060).
- 152 Ellagic acid (4): yellow powder from EtOAc; $C_{14}H_6O_8$.
- 153 β -sitosterol-3-O- β -D-glucopryranoside (5): white amorphous powder from EtOAcC₃₅H₆₀O₆.
- 154 Luteolin-8-*C*-glucoside (6): yellow amorphous powder from $EtOAcC_{21}H_{20}O_{11}$.
- 155 $28-O-\beta$ -D-xylopyranosyl- $(1\rightarrow 4)-\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 2)-[\alpha$ -L-arabinopyranosyl-
- 156 $(1\rightarrow 3)$]-4-*O*-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)- β -D-fucopyranosyl zanhic
- acid (7): white amorphous solid from EtOAc; C₆₀H₉₄O₂₇; HRESIMS (positive-ion mode)
- 158 m/z: 1269.5870 $[M + Na]^+$ (calcd. for C₆₀H₉₄O₂₇Na : 1269.5880).

- 159 $3-O-\beta$ -D-glucopyranosyl-28- $O-\beta$ -D-xylopyranosyl- $(1\rightarrow 4)-\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 2)-4-O$ -
- 160 (3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)- β -D-fucopyranosyl medicagenic acid (8):
- 161 white amorphous solid from EtOAc; $C_{61}H_{96}O_{27}$; HRESIMS (positive-ion mode) m/z:
- 162 1283.6044 $[M + Na]^+$ (calcd. for C₆₁H₉₆O₂₇Na : 1283.6037).
- 163 $3-O-\beta$ -D-glucopyranosyl-28- $O-\beta$ -D-xylopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl(1 \rightarrow 2)-[α -L-
- 164 arabinopyranosyl-(\rightarrow 3)]-4-*O*-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)- β -D-
- 165 fucopyranosyl zanhic acid (9): white amorphous solid from EtOAc; $C_{66}H_{104}O_{32}$; HRESIMS
- 166 (positive-ion mode) m/z: $1431.6395[M+Na]^+$ (calcd. for $C_{66}H_{104}O_{32}Na: 1431.6408$).
- 167 $3-O-\beta-D-glucopyranosyl-(1\rightarrow 4)-\beta-D-glucopyranosyl-28-O-\beta-D-xylopyranosyl-(1\rightarrow 4)-\alpha-L-$
- 168 rhamnopyranosyl- $(1\rightarrow 2)$ - $[\alpha$ -L- arabinopyranosyl- $(1\rightarrow 3)$]-4-O-(3'-hydroxybutanoyloxy-3-169 hydroxybutanoyloxy)- β -D-fucopyranosyl zanhic acid (**10**): white amorphous solid from 170 EtOAc; C₇₂H₁₁₄O₃₇; HRESIMS (positive-ion mode) m/z: 1593.6927[M + Na]⁺(calcd. for 171 C₇₂H₁₁₄O₃₇Na : 1593.6937).
- 172

173 2.4. Antibacterial assay

174 2.4.1. Microorganisms

A total of six bacterial strains were tested for their susceptibility to compounds and 175 these strains were taken from our laboratory collection (kindly provided by Dr. T. 176 177 Ramamurthy, NICED, Kolkata). Among the clinical strains of Vibrio cholerae used in this study, strains NB2 and SG24(1) belonged to O1 and O139 serotypes, respectively. These 178 strains were able to produce cholera toxin and hemolysin (24,25). The other strains used in 179 this study were V. cholerae non-O1, non-O139 (strains CO6 and PC2) (24); and Shigella 180 flexneri (26). The V. cholerae non-O1 and non-O139 strains, were positive for hemolysin 181 production but negative for cholera toxin production (24). The American Type Culture 182 Collection (ATCC) strain, Staphylococcus aureus ATCC 25923, was used for quality control. 183

The bacterial strains were maintained on agar slant at 4 °C and subcultured on a fresh appropriate agar plates 24 h prior to any antibacterial test. The Mueller Hinton Agar (MHA) was used for the activation of bacteria. The Mueller Hinton Broth (MHB) and nutrient agar (Hi-Media) were used for the MIC and MBC determinations respectively.

188

189 2.4.2. Determination of minimum inhibitory concentration (MIC) and minimum bactericidal
190 concentration (MBC)

MIC and MBC of extracts/compounds were assessed using the broth microdilution 191 method recommended by the National Committee for Clinical Laboratory Standards (27,28) 192 with slight modifications. Each test sample was dissolved in dimethylsulfoxide (DMSO, 193 Fisher chemicals) to give a stock solution. The 96-well round bottom sterile plates were 194 prepared by dispensing 180 μ l of the inoculated broth (1x10⁶ CFU/mL) into each well. A 20 195 196 µL aliquot of the compounds was added. The concentrations of tested samples varied from 0.125 to 1024 μ g/mL. The final concentration of DMSO in each well was <1% [preliminary 197 analyses with 1% (v/v) DMSO did not inhibit the growth of the test organisms]. Dilutions of 198 ampicillin (Sigma-Aldrich, Steinham, Germany) and tetracycline (Sigma-Aldrich, Steinham, 199 Germany) served as positive controls, while broth with 20 µL of DMSO was used as negative 200 control. Plates were covered and incubated for 24 h at 37 °C. After incubation, minimum 201 inhibitory concentrations (MIC) were read visually; all wells were plated to nutrient agar 202 (Conda, Madrid, Spain) and incubated for 24 h at 37 °C. The lowest concentrations that 203 yielded no growth after this subculturing were taken as the minimum bactericidal 204 concentration (MBC) values. 205

206

207 2.5. Antioxidant assay

208 2.5.1. DPPH free radical scavenging assay

The free radical scavenging activity of extracts as well as their isolated compounds 209 was performed according to Brand-Williams et al. (29) with slight modifications. Briefly, 210 different concentrations (10 to 2000 µg/mL) of extracts/compounds and vitamin C (Sigma-211 Aldrich, Steinham, Germany) were thoroughly mixed with 3 mL of methanolic DPPH 212 solution (20 mg/L) in test-tubes and the resulting solution was kept standing for 30 minutes at 213 room temperature before the optical density (OD) was measured at 517 nm. The measurement 214 was repeated with three sets and an average of the reading was considered. The percentage 215 216 radical scavenging activity was calculated from the following formula: % scavenging [DPPH] = $[(A_0 - A_1)/A_0] \times 100$. Where A_0 was the absorbance of the control and A_1 was the 217 absorbance in the presence of the samples. IC₅₀ value was determined from the graph obtained 218 using standard vitamin C by using the "y = mx + c" formula from the slope of the graph. 219

220

221 2.5.2. Gallic acid equivalent antioxidant capacity (GAEAC) assay

The GAEAC test was done as previously described (30) with slight modifications. In a 222 223 quartz cuvette, to 950 µL acetate buffer (pH =5.0, 100 mM), the following were added: 20 µL 224 laccase (1 mM stock solution), 20 µL test sample, 10 µL ABTS (2,2'-azinobis(3ethylbenzothiazoline-6-sulfonic acid) (74 mM stock solution). The laccase were purified from 225 Sclerotinia sclerotiorum according to the protocol described (31). The sample concentrations 226 in the assay mixture were 800, 400, 200, 100, 10 µg/mL for the extracts and 200, 100, 50, 25, 227 125.5 µg/mL for the isolated compounds. The content of the generated ABTS^{•+} radical was 228 measured at 420 nm after 240 s reaction time and was converted to gallic acid equivalents by 229 the use of a calibration curve (Pearson's correlation coefficient: r = 0.996) constructed with 0, 230 4, 10, 14, 28, 56, 84 µM gallic acid standards rather than Trolox. Experiments were done in 231 triplicate. 232

234 2.6. Statistical analysis

Data were analyzed by one-way analysis of variance followed by Waller-Duncan Post Hoc test. The experimental results were expressed as the mean \pm Standard Deviation (SD). Differences between groups were considered significant when p < 0.05. All analyses were performed using the Statistical Package for Social Sciences (SPSS, version 12.0) software.

239

240 3. Results and discussion

241 *3.1. Chemical analysis*

According to the antibacterial assays from MeOH, EtOAc and n-BuOH extracts, the 242 EtOAc and *n*-BuOH extracts were submitted to further separation and purification. This led to 243 the isolation of ten compounds. Structures (Figure 1) of these compounds have been assigned 244 on the basis of spectroscopic data (¹H and ¹³C NMR, ¹H-¹H COSY, HSQC, HMBC, ROESY 245 246 and NOESY), mass spectrometry, and by comparison to their data with those of the literature. Hence, the isolated compounds were identified as oleanolic acid (1) (32); 2β -247 248 hydroxyoleanolic acid (2) (32); (2R,3S,2''S)-3''',4',4''',5,5'',7,7''-heptahydroxy-3,8''-249 biflavanone (3) (33); ellagic acid (4) (34); $3-O-\beta$ -D-glucopryranosyl- β - sitosterol (5) (35); luteolin-8-C-glucoside (6) (36); 28-O- β -D-xylopyranosyl-(1 \rightarrow 4)- α -L-rhamnopyranosyl-250 $(1\rightarrow 2)$ - $[\alpha$ -L-arabinopyranosyl- $(1\rightarrow 3)$]-4-O-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)-251 β -D-fucopyranosyl zanhic acid (7) (21); 3-O- β -D-glucopyranosyl-28-O- β -D-xylopyranosyl-252 $(1\rightarrow 4)$ - α -L-rhamnopyranosyl- $(1\rightarrow 2)$ -4-O-(3'-hydroxybutanoyloxy-3-hydroxybutanoyloxy)-253 254 β -D-fucopyranosyl medicagenic acid (8) (21); $3-O-\beta$ -D-glucopyranosyl-28- $O-\beta$ -Dxylopyranosyl- $(1\rightarrow 4)$ - α -L-rhamnopyranosyl $(1\rightarrow 2)$ - $[\alpha$ -L-arabinopyranosyl- $(\rightarrow 3)$]-4-O-(3'-255 hydroxybutanoyloxy-3-hydroxybutanoyloxy)- β -D-fucopyranosyl zanhic acid (9) (20); 3-O-256 β -D-glucopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl-28-*O*- β -D-xylopyranosyl-(1 \rightarrow 4)- α -L-257

258 rhamnopyranosyl-(1→2)-[α-L- arabinopyranosyl-(1→3)]-4-*O*-(3'-hydroxybutanoyloxy-3-259 hydroxybutanoyloxy)-β-D-fucopyranosyl zanhic acid (**10**) (21).

260

261 *3.2. Antibacterial activity*

The susceptibility pattern and inhibition parameters of the tested organisms to the 262 extracts and isolated compounds are indicated below (Table 1). The wells containing a 263 concentration of 64-512 µg/mL of MeOH, EtOAc and *n*-BuOH extracts inhibited the visible 264 growth of all the bacterial species. The most sensitive bacterial species were S. aureus and S. 265 flexneri, while V. cholerae SG24 (1) and V. cholerae NB2 were the most resistant species to 266 267 tested samples. All the three plant extracts showed less antibacterial activity when compared with tetracycline. However, these extracts were active against V. cholerae NB2, V. 268 cholerae PC2 and S. flexneri which were not sensitive to ampicillin. The antimicrobial 269 270 activity of plant extract was considered to be good if its MIC was less than 100.0 µg/mL, moderate if MIC was from 100.0 to 500.0 µg/mL and poor over 500.0 µg/mL (37). Hence, the 271 272 MeOH, EtOAc and n-BuOH extracts of L. leptocarpa exhibited good activity at a MIC value of 64 µg/mL against S. aureus whereas only the MeOH extract displayed poor activity against 273 V. cholerae SG24 (1). The results of the L. leptocarpa extracts showed that this plant species 274 is potential source of antibacterial agents. This in vitro study corroborated the previous 275 276 antibacterial activities of alcoholic extracts from L. octovalvis, L. abyssinica and L. decurrens leaves against Staphylococcus aureus (22,23,38). 277

The lowest MIC and MBC values of 2 μ g/mL were recorded on *S. aureus* with compound **3**; highlighting its good antibacterial potential, as the activity on *S. aureus* was higher than that of ampicillin (MIC = 16 μ g/mL and MBC = 16 μ g/mL) and tetracycline (MIC = 16 μ g/mL and MBC = 128 μ g/mL) used as reference antibacterial drugs. However, the highest MIC value of 512 μ g/mL was recorded on *V. cholerae* SG24 (1) with MeOH extract, and the highest MBC value of 512 μ g/mL was obtained on *V. cholerae* SG24 (1), *V. cholerae* CO6 and *V. cholerae* PC2 with the MeOH extract. A lower MBC/MIC (\leq 4) value signifies that a minimum amount of plant extract/ isolated compound is used to kill the bacterial species, whereas, a higher value signifies the use of comparatively higher concentration of the compounds is needed for the control of the microorganism (*39*).

The antibacterial activities of isolated compounds from L. leptocarpa are in the order 288 289 as compound 3 > compound 6 > compound 2 > compound 4 > compounds 8, 9 > compound 290 10 > compound 7 > compound 1. Compounds 3, 6, 2, 4, 8, 9 and 10 were active against all the tested pathogens whereas compound 1 was active only on S. flexneri and S. aureus. No 291 activity was noted on compound 5 (result not shown). Antimicrobial cut-off points have been 292 defined by several authors to enable the understanding of the potential of pure compounds as 293 follows: significant activity (MIC < 10 μ g/mL), moderate activity (10 < MIC \leq 100 μ g/mL), 294 295 and low activity (MIC> 100 µg/mL) (40,41). Based on this, the antibacterial activity of compound 3 on V. cholerae CO6, V. cholerae NB2, V. cholerae PC2, S. flexneri and S. aureus 296 297 as well as that of compound 6 on Shigella flexneri SDINT and Staphylococcus aureus ATCC 25923 can be considered significant. The strains of V. cholerae NB2, PC2 (24,25) and 298 Shigella flexneri (26) included in the present study were MDR clinical isolates and these were 299 300 resistant to commonly used drugs such as ampicillin, streptomycin, tetracycline, nalidixic acid, furazolidone, co-trimoxazole, etc. However, most of the tested samples displayed 301 antibacterial activities against these microbial strains; suggesting that their administration may 302 represent an alternative treatment against the V. cholerae, the causative agent of dreadful 303 disease cholera and S. flexneri, the causative agent of shigellosis. Taking into account the 304 medical importance of the tested bacteria, this result can be considered as promising in the 305 perspective of new antibacterial drugs development. The antibacterial activities of oleanolic 306 acid, ellagic acid and 2β -hydroxyoleanolic acid corroborate those of the early reports (42,43). 307

All the compounds found to be active in the present study belong to the triterpenoid, flavonoid and phenolic acid groups. Although triterpenoid, flavonoid and phenolic acid compounds have been reported to possess antibacterial activity (*39,44*), no study has been reported on the activity of compounds **3**, **6-10** against these types of MDR pathogenic bacterial strains.

The mechanism of action of terpenoids (1, 2, 5, 7-10) is not fully understood, but is speculated to involve membrane disruption by the lipophilic compounds (45). The inhibition of tested bacterial strains by phenolic acid (4) may be due to iron deprivation or hydrogen bounding with vital proteins such as microbial enzymes (46). The mechanism of the active flavonoids (3, 6) is still to be studied; nevertheless, their activity is probably due to their ability to complex with extracellular and soluble proteins and to complex with bacterial cell walls. More lipophilic flavonoids may also disrupt microbial membranes (47).

320

321 *3.3. Antioxidant activity*

The MeOH, EtOAc and n-BuOH extracts and their isolated compounds were subjected 322 323 for the evaluation of antioxidant activity by using two in vitro model systems. The results were expressed as gallic acid equivalent antioxidant capacity of tested samples (Figure 2) and 324 as equivalent concentrations of test samples scavenging 50% of DPPH radical (Figure 3). 325 DPPH and ABTS+ radical scavenging activities were observed in all the extracts. The 326 MeOH and EtOAc extracts showed dominant activity followed by *n*-BuOH extract (Figures 2 327 and 3) among the extracts. The results indicate the potential of the tried extracts as a source of 328 natural antioxidants with potential application to reduce oxidative stress with consequent 329 health benefits. The antioxidant capacity of tried extracts may be due to the hydrogen 330 donating ability of phenols and flavonoids present in them. Similarly, early reports have 331 shown phenolic compounds to contribute significantly to the antioxidant activity of medicinal 332

plants (39,48). The compounds, which showed the strongest DPPH[•] and ABTS^{•+} radical 333 334 scavenging activities, are compounds 2 (EC₅₀ = 7.66 μ g/mL; GAEAC= 71.64 μ g/mL), 3 $(EC_{50} = 1.09 \ \mu g/mL; \text{ GAEAC} = 96.88 \ \mu g/mL)$ and **6** $(EC_{50} = 10.34 \ \mu g/mL; \text{ GAEAC} = 67.35$ 335 μ g/mL), while the others (compounds 4 and 8) show moderate antioxidant properties. 336 Compounds 1, 5, 7, 9 and 10 were found not active in both the two model systems. 337 Compound **3** was the most antioxidant compound and its DPPH[.] radical scavenging activity 338 was equal to that of vitamin C used in the present study as reference antioxidant drug. The 339 above finding suggests that compound 3 is the best candidate to combat diseases associated 340 with oxidative stress. This is very promising in the perspective of antioxidant drug discover 341 342 from plant origin. The antioxidant activities of compounds 2 and 4 are in agreement with those of the literature (42,49). However, the antioxidant activity of the MeOH, EtOAc and n-343 BuOH extracts from L. leptocarpa as well as that of compounds 3, 6 and 8 are presented here 344 345 for the first time.

346

347 4. Conclusion

The results show that the MeOH and EtOAc extracts from *L. leptocarpa* as well as compounds **2**, **3** and **6** possess the largest antibacterial and antioxidant properties and thus the plant has potentials as a source of natural health-giving products, given further investigations.

351

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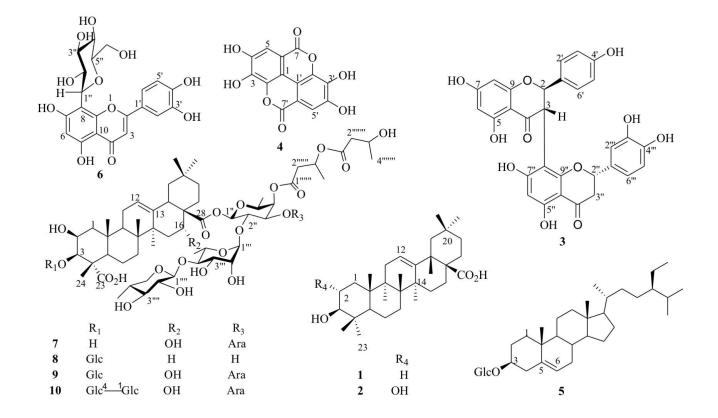
481 Table 1. Antibacterial activity (MIC and MBC in µg/ml) of extracts, isolated compounds

482 and reference antibacterial drugs

Extracts/	Inhibition	Vibrio	Vibrio	Vibrio	Vibrio	Shigella	Staphylococcus
compounds	parameters	cholerae SG24 (1)	cholerae CO6	cholerae NB2	<i>cholerae</i> PC2	<i>flexneri</i> SDINT	aureus ATCC 25923
МеОН	MIC	512	256	256	256	128	64
extract	MBC	512	512	256	512	128	128
entruot	MBC/MIC	1	2	1	2	1	2
EtOAc	MIC	128	256	128	128	128	64
extract	MBC	256	256	>512	256	128	128
	MBC/MIC	2	1	/	2	1	2
<i>n</i> -BuOH	MIC	256	256	128	256	128	64
extract	MBC	256	>512	256	256	128	128
	MBC/MIC	1	/	2	1	1	2
1	MIC	>256	>256	>256	>256	256	256
	MBC	/	/	/	/	>256	>256
	MBC/MIC	/	/	/	/	/	/
2	MIC	128	64	64	64	16	16
	MBC	128	64	64	128	32	16
	MBC/MIC	1	1	1	2	2	1
3	MIC	16	8	8	8	4	2
	MBC	16	8	8	8	4	2
	MBC/MIC	1	1	1	1	1	1
4	MIC	128	64	64	128	64	32
	MBC	>256	128	64	256	64	32
	MBC/MIC	/	2	1	2	1	1
6	MIC	16	32	32	16	4	4
	MBC	32	32	32	16	8	8
	MBC/MIC	2	1	1	1	2	2
7	MIC	>256	256	256	256	128	128
	MBC	/	>256	>256	>256	>256	128
	MBC/MIC	/	/	/	/	/	1
8	MIC	128	256	128	128	128	64
	MBC	>512	256	256	256	128	64
	MBC/MIC	/	1	2	2	1	1
9	MIC	256	128	128	128	64	64
	MBC	>256	256	128	256	128	64

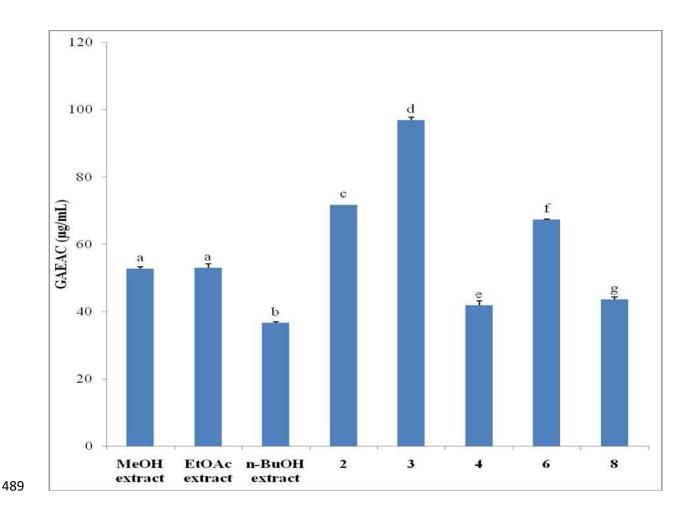
	MBC/MIC	/	2	1	2	2	1
10	MIC MBC MBC/MIC	256 >256 /	256 >256 /	256 >256 /	256 >256 /	128 >256 /	128 128 1
Ampicillin	MIC	16	16	>512	>512	>512	4
	MBC	16	16	>512	>512	>512	4
	MBC/MIC	1	1	/	/	/	1
Tetracycline	MIC	0.5	2	0.5	0.5	16	2
	MBC	4	16	4	4	128	8
	MBC/MIC	8	8	8	8	8	4

483 /: not determined; MIC: Minimum Inhibitory Concentration; MBC Minimum Bactericidal Concentration.
484





488 Figure 1. Structures of compounds isolated from the whole plant of *L. leptocarpa*



490 Figure 2. Gallic acid equivalent antioxidant capacity (GAEAC; μg/ml) of tested samples

Bars represent the mean \pm SD of three independent experiments carried out in triplicate. Letters a-e indicate significant differences between samples according to one way ANOVA and Waller Duncan test; p<0.05. Compounds **1**, **5**, **7**, **9** and **10** were not active (results not shown).

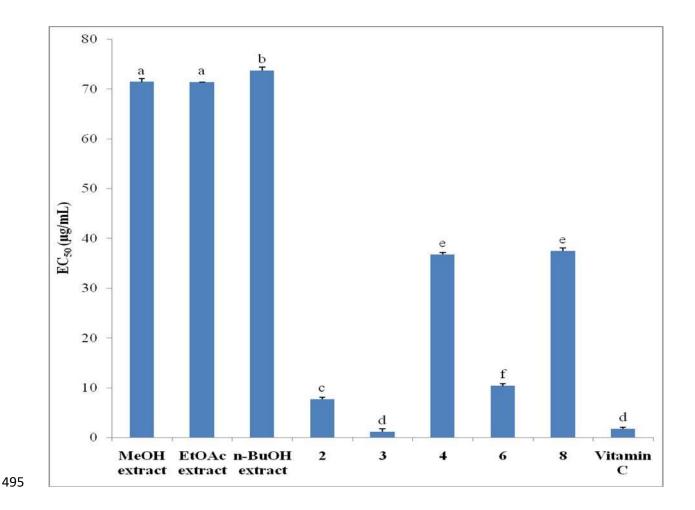


Figure 3. Equivalent concentrations of test samples scavenging 50% of DPPH radical
(EC₅₀). Bars represent the mean ± SD of three independent experiments carried out in
triplicate. Letters a-f indicate significant differences between samples according to one way
ANOVA and Waller Duncan test; p<0.05. Compounds 1, 5, 7, 9 and 10 were not active
(results not shown).