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Perrin Lanversin Foning Tebou, Florence Déclaire Mabou, David Ngnokam, Dominique Harakat, Laurence Voutquenne-Nazabadioko. Owariensisone: a new iridolactone from the whole plant of Brillantaisia owariensis P. Beauv. Natural Product Research, 2016, 30 (14), pp.1611-1616. 10.1080/14786419.2015.1126265. hal-01996505

HAL Id: hal-01996505 https://hal.univ-reims.fr/hal-01996505v1

Submitted on 26 Nov 2021

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Owariensisone: a new iridolactone from the whole plant of *Brillantaisia owariensis* P. Beauv

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ABSTRACT

From the whole plant of *Brillantaisia owariensis* P. Beauv, a new iridolactone, owariensisone (**1**) together with six known compounds (nepetin-7-*O*-glucoside, choline, sucrose, mannitol, xylitol, 1-*O*-palmitoyl-2-eicosanoyl-3-*O*-(6-amino-6-deoxy)- β -D-glucopyranosyl-glycerol) were isolated. Structures of these compounds were established by direct interpretation of their spectral data, mainly HR-TOFESIMS, 1-D NMR (¹H and ¹³C) and 2-D NMR (¹H-¹H COSY, HSQC, HMBC, NOESY, TOCSY and DOCSY) and by comparison with the literature.

KEYWORDS *Brillantaisia owariensis*; Acanthaceae; owariensisone; structure elucidation

1. Introduction

Brillantaisia owariensis P. Beauv (Acanthaceae) is a large erect shrub with purple-blue flowers (Heine, 1963). It is found growing throughout tropical Africa and Madagascar. The leaves of this plant are used in Democratic Republic of Congo folk medicine for the treatment of anaemia (Ngbolua et al. 2013). Previous studies on this species have revealed the presence of sesquiterpene glycoside (Asai et al. 2012). In our continuous search for potentially interesting novel and bioactive secondary metabolites from Cameroonian medicinal plants (Djouossi et al. 2014; Mabou et al. 2014), we have examined the methanol extract of the whole plant

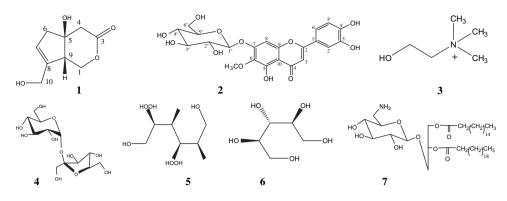


Figure 1. Structures of the isolated compounds (1-7).

of *Brillantaisias owariensis*. In the present paper, we report the isolation of new iridolactone, owariensisone (1) together with six known compounds (Figure 1), from the crude methanol extract of the whole plant.

2. Results and discussion

Compound **1** was obtained as a yellowish gum. Its HR-TOFESIMS exhibited a pseudomolecular ion peak at m/z 207.0631 [M + Na]⁺ (Calcd for C₉H₁₂O₄Na 207.0633) consistent with a molecular formula C₉H₁₂O₄ and indicating four degrees of unsaturation.

The ¹H-NMR spectrum showed one olefinic proton at $\delta_{\rm H}$ 5.68 (H-7, brq^t d, J = 1.8 Hz), two methylenic protons at $\delta_{\rm H}$ 2.59 and 2.64 (H-6a and H-6b), and $\delta_{\rm H}$ 2.79 and 2.82 (H-4a and H-4b), two oxymethylenic protons at $\delta_{\rm H}$ 4.12 (2H, H-10, brs) and $\delta_{\rm H}$ 4.24 and 4.58 (H-1 α and H-1 β) and one methinic proton at $\delta_{\rm H}$ 3.02 (H-9, t, J = 5.3 Hz) attributed to the boschnialactone type skeleton (Sakan et al. 1967; Sisido et al. 1968; Callant et al. 1983; Bianco 1990; Tanaka et al. 1993; Hilgraf et al. 2012). This was supported by its ¹³C-NMR spectrum exhibiting nine carbon signals including one ester carbonyl at $\delta_{\rm C}$ 174.6 (C-3), two ethylenic carbons at $\delta_{\rm C}$ 141.7 (C-8) and 126.9 (C-7), four methylene carbons at $\delta_{\rm C}$ 68.6 (C-1), 60.5 (C-10), 48.0 (C-6) and 44.5 (C-4), one methine carbon at $\delta_{\rm C}$ 55.1 (C-9) and one quaternary carbon at $\delta_{\rm C}$ 80.0 (C-4) bearing an hydroxy group. The ¹J_{C-H} correlation in the HSQC spectrum allowed us to attribute to each carbon the corresponding proton. Thus, the two methylenic protons at $\delta_{\rm H}$ 2.59 and 2.64 (H-6) and 2.79 and 2.82 (H-4) were linked to carbon C-6 and C-4 on the one hand, and the two oxymethylenic protons at $\delta_{\rm H}$ 4.24 and 4.58 (H-1) and 4.12 (H-10) were linked to carbon C-1 and C-10 on the another.

In the COSY spectrum, protons at $\delta_{\rm H}$ 2.59 and 2.64 (H-6a and H-6b) were correlated with proton at $\delta_{\rm H}$ 5.68 (H-7) as well as protons at $\delta_{\rm H}$ 4.24 and 4.58 (H-1a and H-1b) and proton at $\delta_{\rm H}$ 3.02 (H-9, t, J = 5.3 Hz).

From the HMBC spectrum, correlation was observed between the proton at $\delta_{\rm H}$ 4.12 (H-10) and carbon at $\delta_{\rm C}$ 141.7 (C-8), and 126.9 (C-7) suggesting that the oxymethylene was located at C-8. This spectrum also exhibited cross peaks between a proton at $\delta_{\rm H}$ 5.68 (H-7) with carbon at $\delta_{\rm C}$ 55.1 (C-9), 80.0 (C-5) and 48.0 (C-6), while the protons at $\delta_{\rm H}$ 2.79 and 2.82 (H-4) showed correlations with carbons at $\delta_{\rm C}$ 174.6 (C-3), 80.0 (C-5), 55.1 (C-9) and 48.0 (C-6).

The crucial step in the biosynthesis of all cyclic terpenes is the cyclisation step that gives rise to the individual ring systems. In all kingdoms of life, this step has

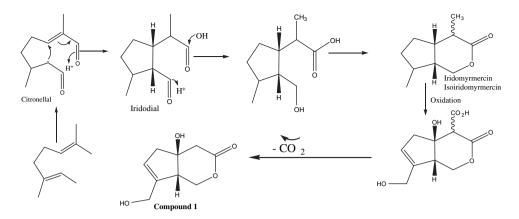


Figure 2. Tentative biosynthetic pathway for compound 1.

been shown to be catalysed by terpene cyclases, which are terpene synthases that use polyprenyl diphosphates cyclases, such as geranyl diphosphate to produce a cationic species that is subsequently cyclised and rearranged to form one of hundreds of possible ring structures (Degenhardt et al. 2009; Chen et al. 2011). However, the cyclisation step that leads to the characteristic 5–6 bicyclic scaffold of iridoids is markedly different. Feeding studies have suggested that the direct precursor of all iridoids in plants is not geranyl diphosphate, but the linear monoterpene 10-oxogeranial (Inouye et al. 1978; Uesato et al. 1983; Uesato, Matsuda, Iida, et al. 1984; Uesato, Matsuda, Inouye 1984). Recently, Fernando and collaborators (2012) reported the discovery of iridoid synthase, a plant-derived enzyme that generates the iridoids ring scaffold; their results illustrate how a short-chain reductase was recruited as cyclase for the production of iridoids in medicinal plants. Therefore, the beta orientation of hydrogen at C-9 was supported by the biosynthetic pathway to iridomyrmercin and isoiridomyrmercin (Lunn 1961) and was further confirmed by the NOESY spectrum on which correlations were observed between H-9 and H-10 as well as H-7 and H-10. The stereochemistry at the C-5 stereogenic was determined to be beta, due to the correlation observed in the NOESY spectrum between H-9 and the proton of the hydroxyl group linked to this stereogenic carbon. The loss of the methyl group at C-4 undergoes probably through oxidation reaction (Figure 2). On the basis of aforementioned information, the structure of this compound was elucidated as lactone of α -(2-hydromethyl-3-hydroxymethylcyclopent-3-*en*-yl) ethanoic acid named owariensisone (Figure 1).

Compounds **4**, **5** and **6** were isolated as a mixture with the same retention factor (*Rf*) on the TLC plate and, TOCSY and DOCSY experiments were useful for determining their structures as sucrose (**4**) (Okada et al. 2010), mannitol (**5**) (Zhao et al. 2008) and xylitol (**6**) (Wang et al. 2013) (Figure 1). Structures of compounds **2**, **3** and **7** (Figure 1) were also determined by means of spectroscopic data and by comparative analysis of their spectral data with those reported in the literature as known nepetin-7-*O*-gluco-side (**2**) (Mohamed et al. 1983; Kenneth et al. 1988), choline (**3**) (Sarkar et al. 2009) and 1-*O*-palmitoyl-2-eicosanoyl-3-*O*-(6-amino-6-deoxy)- β -D-glucopyranosyl-glycerol (**7**) (He et al. 2006) (Figure 1).

3. Experimental

3.1. General

Optical rotations were measured on a Perkin-Elmer 341 polarimeter. ¹H- and ¹³C-NMR spectra were recorded on a Bruker Avance III 600 spectrometer equipped with a cryo-platform (¹H at 600 MHz and ¹³C at 150 MHz). 2-D NMR experiments were performed using standard Bruker microprograms (Xwin-NMR version 2.1 software). Chemical shifts (δ) are reported in parts per million (ppm) with the solvent signals as reference relative to TMS ($\delta = 0$) as internal standard, while the coupling constants (J values) are given in Hertz (Hz). The IR spectra were recorded with a Shimadzu FT-IR-8400S spectrophotometer. UV spectra were determined as methanol solution with a Cary 50 UV/vis Spectrophotometer. HR-TOFESIMS experiments were performed using a Micromass Q-TOF micro instrument (Manchester, UK) with an electrospray source. The samples were introduced by direct infusion in a solution of MeOH at a rate of 5 μ L min⁻¹. Column chromatography (CC) was performed on silica gel 60 (70–230 mesh, Merck) and gel permeation on Sephadex LH-20, while TLC was carried out on silica gel GF₂₅₄ pre-coated plates with detection accomplished by spraying with 50% H₂SO₄ followed by heating at 100 °C, or by visualising with an UV lamp at 254 and 365 nm.

3.2. Plant material

The whole plant of *Brillantaisia owariensis* was collected at Tsinkop, Menoua Division, West Region of Cameroon, in October 2013. Authentication was done by Mr Victor Nana, a botanist of the Cameroon National Herbarium, Yaoundé, where the voucher specimen (N° 34376/ HNC) has been deposited.

3.3. Extraction and isolation

The dried and powdered material (4 kg) was extracted with methanol (3×15 L, 72 h) in a glass tank at room temperature affording 110 g of crude extract after evaporation of the solvent under vacuum.

A portion (100 g) of this crude extract was extracted with ethyl acetate to obtain after evaporation of solvent 60 g of EtOAc extract. Part of methanol-soluble extract (8 g) was subjected to silica gel CC eluting with gradient mixtures of EtOAc–MeOH (5, 10, 20, 30 and 50%). Many fractions, each 250 mL, were collected and combined on the basis of their TLC profiles to give six sub-fractions noted F1–F6. Sub-fraction F1 (1.1 g) was passed through Sephadex LH-20 CC eluted with methanol and 25 fractions were collected. Fractions [15–25] (102 mg) were combined and purified on silica gel CC eluted with EtOAc–MeOH (98:2) to give compound **1** (8 mg). Sub-fractions [2 (4.2 g) was purified on silica gel CC eluted with EtOAc–MeOH (95:5) and many fractions (50 mL each) were collected. Fractions [21–40] (900 mg) were purified on silica gel CC, eluted with the same solvent system to afford the compounds (**2**) (10 mg) and (**7**) (5 mg). Fraction [50–100] (1.5 g) was purified on silica gel CC eluted with EtOAc–MeOH (15 and 20%) and 35 sub-fractions (each, 25 mL) were collected. The sub-fractions [5–6] (500 mg) were repeatedly purified by CC over silica gel, eluted with EtOAc–MeOH (20%) mixture to give the compound (**3**) (10 mg) and the mixture of compounds (**4**), (**5**) and (**6**) (**8** mg).

3.4. New compound information

3.4.1. Owariensisone

Yellowish gum. $[\alpha]_D^{23}$ +2.6 (*c* 0.35 MeOH). IR (NaCl) ν_{max} (cm⁻¹): 3500, 1740, 1615. ¹H-NMR (600 MHz) δ_{H} : 5.68 (1H, brq^t, 1.8, H-7), 4.58 (1H, dd, 11.8 and 4.8, H-1b), 4.24 (1H, dd, 11.8 and 6.6), 4.12 (2H, brs, H-10), 3.02 (1H, brt, 5.3, H-9), 2.82 (1H, d, 14.5, H-4b), 2.79 (1H, d, 14.5, H-4b), 2.64 (1H, dm, 17.8, H-6b) and 2.59 (1H, dt, 17.8, 1.8, H-6a). ¹³C-NMR (150 MHz) δ_{C} . 174.6 (C-3), 141.7 (C-8), 126.9 (C-7), 80.0 (C-5), 68.6 (C-1), 60.5 (C-10), 55.1 (C-9), 48.0 (C-6) and 44.5 (C-4). HR-TOFESIMS *m/z*: 207.0631 [M + Na]⁺ (Calcd for C₉H₁₂O₄Na 207.0633).

Disclosure statement

No potential conflict of interest was reported by the authors.

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